# IN VITRO ANTIOXIDATIVE ACTIVITY OF SOME VIETNAMESE MEDICINAL PLANTS

#### Nguyen Ngoc Hong

Department of Applied Sciences, Ton Duc Thang University, Viet Nam

#### Abstract:

Biomolecules can be oxidized by free radicals. This oxidative damage has an etiological aging and development of disease like cancer, antherosclerosis and other inflamotory disorder. Using synthetic antioxidant can be carcinogenic therefore there is an interesting in researching for antioxidant of natural origin. We report here the result of screening for antioxidant activity of 20 plants with 60 extracts. The antioxidant activity *invitro* was tested for their free radical scavenging activity in the DPPH (1,1-diphenyl-2-picrylhydrazyl) screening assay. In the DPPH method showed that methanol extract of *Adenosma bracteosum* (IC<sub>50</sub> = 3.4 µg/ml) possessed stronger antioxidant properties than acid ascorbic (IC<sub>50</sub>= 3.46 µg/ml).

### **INTRODUCTION**

Oxidative stress is defined as the state in which the level of toxic reactive oxygen species (ROS) overcomes the endogenous antioxidant defences of the host. This state results in an excess of free radicals, which can react with various components of a living cell such as lipids, protein, and nucleic acids, leading to local injury and eventual organ dysfunction. Such damages have been linked to various degenerative diseases including cancer, cataracts, cardiovascular disease and the aging process itself (Ames et al. 1993, Priscilla and Heather 2000).

The additional use of natural antioxidants is an effective way for the prevention and treatment of these conditions. The use of synthetic antioxidants such as butylated hydroxyanisole (BHA) and butylated hydroxytoluene(BHT), due to their toxicity (Namiki,1990) can be at risk of carcinogenicity, therefore, there is a trend in finding of antioxidants of natural origin (Liu and Wang 2000).

The aim of the present study was to evaluate the antioxidant activity of medicinal plants which were selected based on their ethnomedical use for he treatment of rheumatism, fever, inflammation, tonic, hepatitis and cancer. Selected 60 extracts of 20 medicinal plants were evaluated for antioxidant activity using the ferric reducing/antioxidant power assay (FRAP), the 1,1-diphenyl-2-picryl-hydrazyl (DPPH) free radical scavenging assay

# MATERIALS AND METHODS

### Plant material

Most of samples were freshly collected and carefull dried in shade or oven at not higher than 60°C. *Gentiana scabra, Equisetum debile, Periploca sepium, Tetrapanax papyriferus, Oroxylum indicum* were purchased from local herbal market in District 5, Ho Chi Minh City, VietNam.

### Sample preparation and extraction

Dried plant samples were milled with the grinder to medium size powders. 50 grams of powder were extracted with dichloromethane, methanol 95% and water

successively to obtained dichloromethane, methanol 95% and water extracts, respectively using for antioxidant tests.

#### **Determination of FRAP**

The working FRAP reagent was prepared by mixing 10 volumes of 300 mM acetate buffer, pH 3.6, with 1 volume of 10 mM TPTZ (2,4,6-tripyridyl-s-triazine) in 40mM hydrochloric acid and with 1 volume of 20mM ferric chloride. Freshly prepare FRAP reagent (1.5ml) was warmed to 37°C. Subsequently, 50µl of sample and 150 µl of deionized water was added to the FRAP reagent. Absorbance readings were taken after 1.5 h. Standard curve was prepared using different concentrations (100-1000 µM) of FeSO<sub>4</sub>.7H<sub>2</sub>O. All solutions were used on the day of preparation. The results were corrected for dilution and expressed in  $\mu$ M Fe<sup>2+</sup>/L. Acid ascorbic was measured within 1 h after preparation. FRAP assay measures the change in absorbance at 595 nm owing to the formation of a blue colored Fe<sup>II</sup>-tripyridyltriazine compound from colorless oxidized Fe<sup>III</sup> form by the action of electron donating antioxidants (Benzie and Strain 1996, Pulido et al. 2000). All determinations were performed in triplicate

## Free radical scavenging capacity

The free radical scavenging capacity of samples and pure compound was analyzed by DPPH assay. Aliquots (50  $\mu$ l) of the tested samples were mixed with 2 ml of 6.10<sup>-5</sup> M methanolic solution of DPPH radical. Methanolic solution of pure compounds were tested too. Absorbance measurements commenced immediately. The decrease in absorbance at 515 nm was determined after 30 min for all samples. Methanol was used to zero spectrophotometer .

All determinations were performed in triplicate. The percent inhibition of the DPPH radical by the samples was calculated according to the formula proposed of Yen an Dul (1994):

#### % inhibition = $[(A_{C(0)} - A_{A(t)})/A_{C(0)}]x100$

where  $A_{C(0)}$  is the absorbance of the control at t = 0 min and  $A_{A(t)}$  is the absorbance of the antioxidant at t = 30 min

For determination of IC<sub>50</sub>, ascorbic was used as reference

### **RESULTS AND DISCUSSION**

## Total antioxidative capacity of 60 medicinal plant extracts

The yield of crude extracts (grams of extract/ 100 grams of sample), antioxidative capacity, free radical scavenging activity and in vitro XO inhibition of extracts were shown in table 1. There were big differences in the antioxidative capacity FRAP between the selected medicinal plant extracts. The FRAP values varied from 147 to 4333  $\mu$ mol Fe<sup>2+</sup>/L of 1 mg/ml sample concentrations.

According to their reducing ability/antioxidative power, the antioxidative effect of these thirty six medicinal plant extracts can be divided into four groups: (a) low (< 1 mM), n = 29; (b) average (1-2.5 mM), n = 21; (c) good (2.5-4 mM), n = 7; and (d) high (> 4 mM), n = 3.

As shown in table 1 and Fig.1, the extracts with the strongest antioxidative properties when measured with the FRAP assay (in order) were: *Adenosma bracteosum* (M extract) > *Phyllanthus amarus* (H extract) > *Periploca sepium* (M extract) > *Periploca sepium* (H extract) > *Oroxylum indicum*.(M extract) >

STT	Plant name	Local name	Parts used	Ex	tracted y	ield	FRA	P (jumolł	e <sup>2-</sup> /L)	% DF	PH inhil	oition
2				Q	Μ	M	٩	Μ	M	Q	Μ	W
	Acanthus ilicifolius	Ô rô nước	Leaves, stems	2.34	8.2	20.02	527	671	512	11.48	17.41	13.49
~i	Achyranthes bidentata	Ngưn tất	Whole plant	1.51	7.97	22.1	433	1073	1170	18.63	22.36	52.29
3.	Adenosma bracteosum	Nhân trần Tày Ninh	Whole plant	1.56	10.78	20.36	555	4333	3019	26.7	90.62	36.93
4	Artemisia vulgaris	Ngài cứu	Whole plant	2.01	10.87	17.45	837	2335	3001	11.27	28.38	25.91
5.	Crescentia cujete	Đào tiên	Fruits	4.36.	10.07	19.3	438	1344	725	24.28	31.93	30.53
6	Eclipta prostrate	Cỏ mực, nhọ nồi	Whole plant	0.89	11.84	28.31	916	1429	2316	96.6	42.81	59.2
7.	Eleusine indica	Mần trầu	Whole plant	2.17	8.53	20.56	526	147	1195	13.87	19.51	13.26
8.	Equisetum debile	Mộc tặc	Whole plant	3.27	8.93	23.1	742	1780	1029	35.53	66.11	29.04
9.	Gentiana scabra	Long đởm thảo	Roots	4.95	12.8	23.6	729	1893	1104	64.66	69.23	40.58
10.	Glinus oppositifolius	Rau đắng đất	Whole plant	1.89	8.83	18.9	483	1233	877	18.44	26.24	49.95
11.	Momordica charantia	Mướp đắng	Fruits	1.23	12.9	20.3	378	1079	682	38.98	31.79	26.94
12.	Morinda persicaefolia	Nhàu nước	Roots	0.98	13.6	23.9	634	1347	657	65.27	54.86	50.51
13.	Oroxylum indicum	Núc nác, hoàng bá nam	Stem bark	1.01	9.94	15.3	2527	3139	2575	82.03	68.02	67.83
14.	Oxalis corniculata L.	Chua me đất	Whole plant	3.71	6.04	21.18	811	848	937	12.46	21.8	10.97
15.	Periploca sepium	Hương gia bì	Stem bark	5.95	14.56	21.34	1045	4088	3393	21.24	71.8	36.51
16.	Phyllanthus amarus	Diệp hạ châu	Whole plant	4.36	9.14	20.64	553	2008	4146	27.92	24.18	72.13
17.	Plumbago zeylanica	Bạch hoa xà	Whole plant	3.89	14	22.3	719	2513	1687	15.13	31.14	25.02
18.	Polygonum aviculare	Rau đắng	Whole plant	2.07	8.67	28	660	1690	1894	27.54	29.74	33.75
19.	Scutellaria barbata	Bán chỉ liên	Whole plant	1.51	7.97	22.1	936	2349	1868	44.77	67.41	48.79
20.	Tetrapanax papyriferus	Thòng thảo	Core of stems	0.32	2.5	40.49	824	882	613	13.35	26.61	9.52
	Al	dueviation of extracts. I	D: dicloromethar	le; M: 1	methand	ol: H: ac	lueous	extract	s			

Table 1: Antioxidative activities of 60 plant extracts on FRAP, DPPH' free radical

Hội nghị khoa học lần thứ 1 – Khoa KHƯD

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# Tiểu ban: Công nghệ sinh học



Fig. 1. Total antioxidant capacity determined as FRAP of 60 extracts *Note: Abbreviation of*  $N^0$  *is plant extracts in table 1.* 

Adenosma bracteosum (H extract) > Artemisia vulgaris (H extract)> Oroxylum indicum (H extract) > Oroxylum indicum (D extract) > Plumbago zeylanica (M extract) > Scutellaria barbata (M extract).

In general, values of methanolic or water extracts were stronger than that of dichloromethane ones. Depending on samples, the methanolic or aqueous can be stronger than the other.

Free Radical Scavenging Ability of Medicinal Plant Extracts





 Table 2. IC<sub>50</sub> values of 5 extracts and reference compounds

		IC <sub>50</sub>
Names	Extract	(µg/ml)
Phyllanthus amarus	W	15.55
Oroxylum indicum	D	13.02
Adenosma bracteosum	М	3.4
Periploca sepium	М	15.65
Ascorbic Acid		3.46

As shown in table 1 and fig.2, there were three different kinds of extracts according to their activities: 12 plant samples showed activity of M extracts were stronger than that of D and H extracts. On the other hand, activity of H extracts of other 5 plant samples were stronger than M and D extracts. Finally, 3 plant samples had D extract activity were the strongest ones.

The highest 4 active extracts in DPPH assay among tested ones (in order) were: *Adenosma bracteosum* (M extract) > *Oroxylum indicum* (D extract) > *Phyllanthus amarus* (H extract) > *Periploca sepium* (M extract). These extracts were further determined their IC<sub>50</sub> values in comparison with reference ascorbic acid.

The result was shown in Fig. 3 and Table 2. Each  $IC_{50}$  value was achieved from a linear regression analysis showing good correlation coefficient ( $r^2 \ge 0.9$ ).

IC<sub>50</sub> of crude M extract of *Adenosma bracteosum* which showed IC<sub>50</sub> of 3.4  $\mu$ g/ml has better *in vitro* antioxidative activity than that of ascorbic acid (IC<sub>50</sub>= 3.46  $\mu$ g/ml)

## Comparison of Total Antioxidant FRAP and DPPH Radical Scavenging Properties

The FRAP mechanism is totally electron transfer. Hence, in combination with DPPH assay, the FRAP can be very useful in distinguishing dominate mechanisms with different antioxidants. The DPPH assay is considered to be mainly based on an electron transfer reaction and hydrogen atom abstraction is marginal reaction pathway (Ronald et al. 2005). Some extracts had the different levels of antioxdative capability in each method. This may be explained to be FRAP reaction activity take place in aqueous solution, for this reason, the polarized compounds easily react in this solution, and the orthers meet a prolem when reaction in polarized solvent. In DPPH assay, reaction happen in methanolic solvent, consequently, compounds is not very polarized react well in this solution.

## CONCLUSIONS

The results of the study demonstrate that some medicinal plants are promising sources of natural antioxidants. The strongest antioxidative properties when measured with FRAP, DPPH among sixty test samples were methanolic extracts of *Adenosma bracteosum* and H extract of *Phyllanthus* These extract will be further for isolating of active compounds.

## REFERENCES

- Ames, B.N., Shigenaga, M.K. and Hagen, T.M. 1993. Oxidants, antioxidants, and the degenerative diseases of aging. Proceeding of National Academy of Sciences of United States of America 90, pp. 7915-7922
- Benzie, I. F., and Strain, J. J. 1996. The ferric reducing ability of plasma (FRAP) as measurement of "antioxidant power": The FRAP assay. Analytical Biochemistry 239: 70–76
- Halliwell, B. and Gutteridge, N.-J-C. 1999. Free radicals in Biology and Medicine, Oxford University Press, Oxford, U.K
- Ronald, L. P., Xianli, W., and Karen, S. 2005. Standardized methods for the determination of antioxidant capacity and phenolics in foods and dietary supplements. Agricultural and food chemistry 53 (10): 4290-4302
- V.Katalinic, M.Milos, T. Kulisic, M. Jukic.2006. Screening of 70 medicinal plant extracts for antioxidant capacity and total phenols. Food chemistry 94. 550-557
- Yen, G.C, and Duh, P.D. 1994. Scavenging effect of methanolic extracts of peanut hulls on free radical and active oxygen species. Agricultural and Food Chemistry 42, pp.629-63